

Characterization of lscf cathode material modified with f-cnts

Abstract

Cathode is one of the important parts in performing the high efficiency of proton conducting fuel cell (PCFC). Selection of appropriate cathode material may resolve the major drawbacks at the cathode part associated with the high R_p . Accordingly, tremendous effort have been done to reduce the R_p and one of the alternatives is the modification of cathode microstructure that can be achieved by introducing dispersing agent in the synthesis route. Thus, in this present work, a functionalized carbon nanotubes (f-CNTs) obtained from acidification process was used as a dispersing agent in the synthesis of $\text{La}_{0.6}\text{Sr}_{0.4}\text{Co}_{0.2}\text{Fe}_{0.8}\text{O}_{3-\delta}$ (LSCF) cathode material. The amount of 4 mg, 8 mg and 12 mg of f-CNTs were respectively added to LSCF cathode during the synthesizing process by a sol-gel method. Semi-solid gel obtained was calcined at $900\text{ }^\circ\text{C}$ to form high purity of LSCF powder and respectively denoted as LSCF₄, LSCF₈ and LSCF₁₂. The powder was characterized by Fourier Transform Infrared (FTIR) Spectroscopy, Pycnometer, Particle Size Analyzer and Scanning Electron Microscopy with Energy Dispersive X-ray (SEM/EDX). The FTIR analysis depicted the peak of respective metal complexes, metal oxide, symmetrical and asymmetrical stretching of carboxylate. The pycnometer showed the lowest density of LSCF₄ was 2.8777 g/cm^3 . The Particles Size Analyzer confirmed the particle size of 38 nm ultrafine powder for LSCF₄. The SEM image depicted the highly disperse spherical particles found in LSCF₄ with particle size about 30 nm. The elemental composition of the samples is comparable with the nominal stoichiometric of LSCF₄ as corroborated by the EDX analysis. Therefore, the LSCF with optimum 4 mg f-CNTs as dispersing agent has potential as nanoporous cathode material for proton conductivity fuel cell.

Keywords

Dispersing agent; F-CNTs; LSCF cathode; Sol-gel