

PREPARATION, CHARACTERIZATION AND  
DEGRADATION OF LINEAR LOW DENSITY  
POLYETHYLENE / RAMBUTAN PEEL FLOUR  
BIOCOMPOSITES

AINATUN NADHIRAH BINTI AZMI

1433511236

UNIVERSITI MALAYSIA PERLIS

2016



**PREPARATION, CHARACTERIZATION AND  
DEGRADATION OF LINEAR LOW DENSITY  
POLYETHYLENE / RAMBUTAN PEEL FLOUR  
BIOCOMPOSITES**

**AINATUNNADHIRAH BINTI AZMI**

**(1433511236)**

A thesis submitted in fulfillment of the requirements for the degree of  
Master of Science in Biosystems Engineering

**School of Bioprocess Engineering  
UNIVERSITI MALAYSIA PERLIS**

2016

## ACKNOWLEDGEMENT

First and foremost, my utmost gratitude to my supervisor, Dr. Sam Sung Ting whose help me a lot in my journey to complete my research and thesis. His excellent in guidance, caring, patiently corrects my writing day and nights will not be forgettable. I really appreciate his enthusiasm and ideas towards this research.

Thank you does not seem sufficient but it said with appreciation and respect. Special appreciation goes to my co-supervisor Dr. Nik Noriman Bin Zulkepli for his supervision and knowledge to this research.

I would like to express my indeed grateful to my beloved parents, Mr. Azmi Bin Ahmad and Mrs. Halimaton Sakdiah Bt Abu Bakar for their endless love, encouragement and prayers that make this research become possible.

I would also like to thank to all teaching engineers and technicians who rendered their helps during the period of my project work.

Last but not least, sincere thanks to all my colleagues for sharing their knowledge and their moral support. I owed all of u a lot.

## TABLE OF CONTENTS

	<b>PAGE</b>
<b>THESIS DECLARATION</b>	i
<b>ACKNOWLEDGEMENT</b>	ii
<b>TABLE OF CONTENTS</b>	iii
<b>LIST OF TABLES</b>	vii
<b>LIST OF FIGURES</b>	ix
<b>LIST OF ABBREVIATIONS</b>	xii
<b>LIST OF SYMBOLS</b>	xiv
<b>ABSTRAK</b>	xv
<b>ABSTRACT</b>	xvi
<b>CHAPTER 1 : INTRODUCTION</b>	
1.1 Overview	1
1.2 Problem Statement	5
1.3 Objectives	6
1.4 Scope of Study	6
1.5 Thesis Organization	6
<b>CHAPTER 2 : LITERATURE REVIEW</b>	
2.1 Introduction to Solid Waste Issue	8
2.2 Polyethylene	10
2.2.1 Low Density Polyethylene (LDPE)	11
2.2.2 Linear Low Density Polyethylene (LLDPE)	12
2.3 Degradable Polymer	13
2.3.1 Natural Polymer	14
2.3.1 (a) Starch	14

2.3.2 (b)	Cellulose	16
2.3.3 (c)	Chitin and Chitosan	18
2.3.2	Synthetic polymer	19
2.4	Degradation of Polyolefin	20
2.4.1	Biodegradable Polyolefin	21
2.4.2	Photo-degradation	21
2.4.3	Thermo-oxidative Degradation	24
2.5	Degradation Condition	25
2.5.1	Natural Weathering	25
2.5.2	Soil Burial	26
2.6	Oxo-degradable Polymer	27
2.7	Rambutan Peel	28
2.7.1	Rambutan Waste	28
2.7.2	Composition of Rambutan Peel	29
2.7	Compatibilization in Polymer Biocomposites	29
2.8	Pro-oxidant in Plastics	30
<b>CHAPTER 3 : METHODOLOGY</b>		
3.1	Introduction	32
3.2	Materials	32
3.3	Preparation of Samples	35
3.3.1	Preparation of Rambutan Peel Flour (RPF)	35
3.3.2	Compatibilizer (RPF-AA) Preparation	35
3.3.3	Composites Preparation	35
3.4	Analysis	37
3.4.1	Tensile Properties	37
3.4.2	Morphological Study	37
3.4.3	Fourier Transform Infrared (FTIR) Analysis	37
3.4.4	Differential scanning Calorimetry (DSC)	38

3.4.5	Thermogravimetric Analysis (TGA)	39
3.4.6	Water Absorption	39
3.5	Degradation Test	39
3.5.1	Natural Soil Burial Test	40
3.5.2	Natural Weathering Test	41

## **CHAPTER 4 : RESULTS AND DISCUSSION**

4.1	Adipic Acid as Compatibilizer	43
4.1.1	Compatibilization of Adipic Acid on LLDPE/RPF Composites	43
4.1.2	Tensile Properties	46
4.1.3	Differential Scanning Calorimeter (DSC)	52
4.1.4	Thermogravimetric Analysis (TGA)	55
4.1.5	Water Absorption	57
4.2	Effect of Natural Weathering on the Adipic Acid Compatibilized LLDPE/RPF Composites	59
4.2.1	Tensile Properties	59
4.2.2	Carbonyl Indices	67
4.2.3	Differential Scanning Calorimeter	69
4.3	Effect of Natural Soil Burial on the Adipic Acid Compatibilised LLDPE/RPF Composites	71
4.3.1	Tensile Properties	71
4.3.2	Carbonyl Indices	76
4.3.3	Differential Scanning Calorimeter	79
4.4	Incorporation of Cobalt Stearate in LLDPE/RPF-AA Composites	82
4.4.1	Natural Weathering	82
4.4.1(a)	FTIR and Carbonyl Indices	82
4.4.1(b)	Tensile Properties	85
4.4.1(c)	Differential Scanning Calorimeter	90

4.4.2	Natural Soil Burial	92
4.4.2(a)	Tensile Properties	92
4.4.2(b)	Carbonyl Indices	97
4.4.2(c)	Differential Scanning Calorimeter	98
<b>CHAPTER 5 : CONCLUSIONS AND RECOMMENDATION FOR FUTURE STUDY</b>		
5.1	Conclusions	100
5.2	Recommendations for Future Study	101
<b>REFERENCES</b>		103
<b>APPENDIX</b>		115
<b>LIST OF PUBLICATIONS</b>		119

© This item is protected by original copyright

## LIST OF TABLES

NO.		PAGE
1.1	Generation of total solid waste in Peninsular Malaysia	1
3.1	Raw materials, function and supplier	33
3.2	Purposes, instruments and model	33
3.3	Table of formulation for series 1 (Effect of different loading to LLDPE/RPF composites)	36
3.4	Table of formulation for series 2 (Effect of adipic acid as compatibilizer on the LLDPE/RPF composites)	36
3.5	Table of formulation for series 3 (Effect of cobalt stearate as pro-oxidant on the LLDPE/RPF-AA composites)	36
3.6	Mean temperature, rainfall and relative humidity for the Perlis state from nearest meteorology station in Chuping, Perlis	41
4.1	Thermal properties analysis for LLDPE, LLDPE/RPF composites and LLDPE/RPF-AA composites	53
4.2	Weight loss (%) for LLDPE, LLDPE/RPF composites and LLDPE/RPF-AA composites	56
4.3	Data of tensile strength (MPa), elongation at break (%) and Young's modulus (MPa) for LLDPE/RPF composites and LLDPE/RPF-AA composites up to 9 months of natural weathering	62
4.4	Carbonyl index for LLDPE/RPF composites and LLDPE/RPF-AA composites up to 9 months of natural weathering	69
4.5	DSC results of LLDPE, LLDPE/RPF composites and LLDPE/RPF-AA composites after different period of natural weathering	70
4.6	Data of tensile strength (MPa), elongation at break (%) and Young's modulus (MPa) for LLDPE/RPF composites and LLDPE/RPF-AA composites up to 9 months of soil burial	73

4.7	DSC results of LLDPE, LLDPE/RPF composites and LLDPE/RPF-AA composites after different period of soil burial	81
4.8	Carbonyl index for LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites up to 9 months of natural weathering.	85
4.9	Data of tensile strength (MPa), elongation at break (%) and Young's modulus (MPa) for LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites up to 9 months of natural weathering	88
4.10	DSC results of LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites after different period of natural weathering	91
4.11	DSC results of LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites after different period of natural soil burial	99

© This item is protected by original copyright

## LIST OF FIGURES

NO.		PAGE
1.1	Component of solid waste (%)	2
2.1	Typical structure of LLDPE	13
2.2	Structure of amylose and amylopectin	15
2.3	Chemical structure of chitin	18
2.4	Typical photo-degradation process of polymer	23
2.5	Degradation steps of thermal degradation	25
3.1	Overall process for natural weathering and soil burial test	34
3.2	Samples buried in the soil	40
3.2	Samples on the natural weathering rack	42
4.1	Formation of ester bond between RPF and adipic acid	44
4.2	Schematic diagram of (a) Structure of RPF and LLDPE before compatibilized and (b) Homogeneous structure of RPF and LLDPE compatibilized by adipic acid.	45
4.3	IR spectra of pure LLDPE, LLDPE/25% RPF composites and LLDPE/25% RPF-AA composites	46
4.4	Tensile strength of LLDPE/RPF composites and LLDPE/RPF-AA composites in various RPF loading	47
4.5	Elongation at break ( $E_b$ ) of LLDPE/RPF composites and LLDPE/RPF-AA composites in different RPF loading.	48
4.6	Young's modulus LLDPE/RPF composites and LLDPE/RPF-AA composites in various RPF loading	49
4.7	Tensile fracture surface (x500) of LLDPE/RPF composites with RPF loading of (a) 0 wt% (b) 5 wt% (c) 15 wt% and (d) 25 wt%.	51
4.8	Tensile fracture surfaces (x500) of LLDPE/RPF-AA composites RPF-AA loading of (a) 5 wt%, (b) 15 wt%, and (c) 25 wt%.	51

4.9	DSC melting thermogram of LLDPE, LLDPE/RPF composites and LLDPE/RPF-AA composites	52
4.10	DSC heating thermogram of LLDPE, LLDPE/RPF composites and LLDPE/RPF-AA composites	53
4.11	Weight loss versus temperature of LLDPE, LLDPE/RPF composites and LLDPE/ RPF-AA composites	55
4.12	Water absorption for LLDPE/RPF composites and LLDPE/RPF-AA composites	58
4.13	SEM micrographs (x500) of LLDPE/RPF composites with RPF loading of (a) 0 wt% (b) 5 wt% (c) 25 wt% after 3 months; (d) 0 wt% (e) 5 wt% (f) 25 wt% after 6 months; (g) 0 wt% (h) 5 wt% (i) 25 wt% after 9 months of natural weathering	65
4.14	SEM micrographs (x500) of LLDPE/RPF-AA composites with RPF-AA loading of (a) 5 wt% (b) 25 wt% after 3 months; (c) 5 wt% (d) 25 wt% after 6 months; (e) 5 wt% (f) 25 wt% after 9 months of natural weathering	66
4.15	IR spectra of LLDPE/RPF composites over 9 months of weathering	67
4.16	IR spectra of LLDPE/RPF-AA composites over 9 months of weathering	68
4.17	SEM micrographs (x500) of LLDPE/RPF composites with RPF loading of (a) 0 wt% (b) 5 wt% (c) 25 wt% after 3 months; (d) 0 wt% (e) 5 wt% (f) 25 wt% after 6 months; (g) 0 wt% (h) 5 wt% (i) 25 wt% after 9 months of soil burial	74
4.18	SEM micrographs (x500) of LLDPE/RPF-AA composites with RPF-AA loading of (a) 5 wt% (b) 25 wt% after 3 months; (c) 5 wt% (d) 25 wt% after 6 months; (e) 5 wt% (f) 25 wt% after 9 months of soil burial	75
4.19	IR spectra of LLDPE/RPF composites over 9 months composting period	77
4.20	IR spectra of LLDPE/RPF-AA composites over 9 months composting period	78
4.21	Carbonyl index for LLDPE/RPF composites and LLDPE/RPF-AA composites over different soil burial period	79

4.22	Comparison of the IR spectra of LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites after 9 months of natural weathering	83
4.23	Degradations path away of polyethylene containing pro-oxidant	84
4.24	SEM micrograph (x500) of CS-incorporated LLDPE/RPF-AA composites with loading of (a) 5 wt% (b) 25 wt% after 3 months; (b) 5 wt% (c) 25 wt% after 6 months; (e) 5 wt% (f) 25 wt% after 9 months of natural weathering	89
4.25	Tensile strength of LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA up to 9 months of soil burial	93
4.26	$E_b$ of LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites up to 9 months of soil burial	94
4.27	Young's modulus of LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites up to 9 months of soil burial	95
4.28	SEM micrograph (x500) of CS-incorporated LLDPE/RPF-AA composites with loading of (a) 5 wt% (b) 25 wt% after 3 months; (b) 5 wt% (c) 25 wt% after 6 months; (e) 5 wt% (f) 25 wt% after 9 months of soil burial.	96
4.29	Carbonyl indices of LLDPE/RPF-AA composites and CS-incorporated LLDPE/RPF-AA composites up to 9 months of soil burial	98

## LIST OF ABBREVIATIONS

AA	Adipic acid
AAc-g-PLDPE	Acrylic acid grafted LDPE after plasma treatment
CaSt	Calcium stearate
CA	Citric acid
Co	Cobalt
CO <sub>2</sub>	Carbon dioxide
CS	Cobalt stearate
DSC	Differential scanning calorimetry
E <sub>b</sub>	Elongation at break
Fe	Iron
FTIR	Fourier transform infrared
GLU	Glutaraldehyde (GLU)
H <sub>2</sub> O	Water
HDPE	High density polyethylene
HF	Hemp fiber
HMTA	Hexamethylenetetramine
MA-g-PLDPE	Maleic anhydride grafted LDPE after plasma treatment
Mn	Manganese
MSW	Municipal solid waste
NaOH	Sodium hydroxide
LDPE	Low density polyethylene
LLDPE	Linear low density polyethylene

PCL	Polycaprolactone
PE	Polyethylene
PE-g-MA	Polyethylene-grafted-maleic anhydride
PET	Polyethylene terephthalate
PGA	Poly (glycolic acid)
PLA	Poly (lactic acid)
PMMA	Polymethylmethacrylate
PP	Polypropylene
PS	Polystyrene
PSW	Plastic solid waste
PVC	Polyvinyl chloride
PVOH	Poly (vinyl alcohol)
RPF	Rambutan peel flour
RWF	Rambutan skin waste flour
SEM	Scanning electron microscopy
TGA	Thermogravimetric analysis
TPS	Thermoplastic starch
TPSS	Thermoplastic sago starch
UTM	Universal Testing Machine
UV	Ultra violet

## LIST OF SYMBOLS

%	percentage
wt %	weight percentage
Mt	metric ton
g	gram
kg	kilogram
mg	milligram
°C	degree celcius
°C/min	degree celcius per minute
nm	nanometer
µm	micrometre
MPa	mega pascal
GPa	giga pascal
h	hour
L	litre
ml	milimetre
g/cm <sup>3</sup>	gram per cubic centimeter
Da	Dalton
kV	kilo volt
$\Delta H^*_{*f}$	heat fusion for 100% crystalline polyethylene
$\Delta H^{\circ}_f$	heat of fusion for semi crystalline polyethylene
cm <sup>-1</sup>	reciprocal wavelength
cm <sup>3</sup>	cubic centimeter
mm <sup>3</sup>	cubic milimetre
mm/min	milimetre per minute

## **Perincian dan Sifat – sifat Terhadap Keupayaan Degradasi oleh Campuran Linear Polietilena Berketumpatan Rendah / Tepung Kulit Rambutan Biokomposit**

### **ABSTRAK**

Pada masa kini, perkembangan plastik dari polimer terbiodegradasi dan pengisi semulajadi telah menarik minat besar dalam bidang sains dan penyelidikan. Polimer terbiodegradasi boleh membenarkan degradasi lengkap di dalam tanah dan tidak mengeluarkan mana-mana komponen toksik atau berbahaya. Oleh itu, dalam kajian ini, percubaan telah dibuat dengan menggabungkan tepung kulit rambutan (RPF) ke dalam matriks linear polietilena berketumpatan rendah (LLDPE). Kajian ini menyiasat pengaruh asid adipik sebagai pengserasi pada sifat-sifat komposit menggabungkan linear polietilena ketumpatan rendah dengan kandungan tepung kulit rambutan bervariasi dari 5-25% kandungan. Kobalt stearat sebagai pro-oksidan dengan berat 0.2% telah ditambah ke dalam komposit LLDPE/RPF-AA. Pendedahan polimer kepada persekitaran semulajadi dan ditanam ke dalam tanah dijalankan untuk tempoh 9 bulan untuk mengenal pasti tahap potensi polimer terdegradasi dalam persekitaran yang berbeza. Hasilnya, komposit LLDPE/RPF-AA menunjukkan kekuatan tegangan dan pemanjangan pada takat putus yang lebih tinggi berbanding komposit LLDPE/RPF selepas pendedahan kepada persekitaran dan ditanam di dalam tanah. Walaubagaimanapun, selepas penambahan kobalt stearat, kekuatan tegangan dan pemanjangan takat putus komposit LLDPE/RPF-AA menjadi berkurangan. Pengimbas elektron mikroskopi menunjukkan mikrograf pembentukan retak, liang dan koloni kulat pada permukaan komposit LLDPE/RPF dan komposit LLDPE/RPF-AA. Untuk analisis pengimbas pembezaan kalorimeter, komposit LLDPE/RPF-AA menghasilkan penghabluran yang lebih tinggi berbanding dengan komposit LLDPE/RPF. Kestabilan haba bagi setiap komposit menurun daripada 429 °C bagi LLDPE dengan pertambahan sebanyak 25% kandungan tepung kulit rambutan (251 °C) dan tepung kulit rambutan bersama asid adipik (262 °C).

## Characterization and Properties on Degradability of Linear Low Density Polyethylene/Rambutan Peel Flour Biocomposites

### ABSTRACT

Nowadays, developments of plastics from biodegradable polymers and natural filler have attracted great interests in science and research. Biodegradable polymer could allow complete degradation in soil and does not emit any toxic or noxious components. Therefore, in this research, an attempt was made by incorporating rambutan peel flour (RPF) into LLDPE matrix. This study investigated the influence of adipic acid as compatibilizer on the properties of the composites prepared from LLDPE with different RPF loading varied from 5 to 25 wt%. Cobalt stearate (CS) as a pro-oxidant (0.2 wt%) has been added into LLDPE/RPF-AA composites. Natural weathering and soil burial test were performed for 9 months' period to determine the potential of this polymer to degrade in different surroundings. As a result, LLDPE/RPF-AA composites showed higher tensile strength and elongation at break ( $E_b$ ) compared to LLDPE/RPF composites after weathering and composting exposure. However, after the addition of CS, the tensile strength and  $E_b$  of LLDPE/RPF-AA composites decreased. Scanning electron microscopy (SEM) micrographs showed the formation of cracks, pores and fungus colonization on the LLDPE/RPF composites and LLDPE/RPF-AA composites surface. For differential scanning calorimetry (DSC) analysis, LLDPE/RPF-AA composites presented higher crystallinity compared to the LLDPE/RPF composites. The thermal stability for the composites decreased from 429 °C (LLDPE) to 251 °C for 25 wt% of RPF and 262 °C for 25 wt% of RPF-AA.

# CHAPTER 1

## INTRODUCTION

### 1.1 Overview

Nowadays, the disposal of plastic waste has been continuously on the increase and its has been a big problem up to today in Malaysia. Most of the solid waste is contributed by petroleum-based plastics such as polyethyele (PE), polypropylene (PP), polyethylene terephtalate (PET), polivinyll chloride (PVC) and polystyrene (PS). The application of the plastics typically used for packaging purposes but also in non-plastics applications such as textile fibres and coatings (Dewil et al., 2006). Apart from that, 40% of the plastic that used in packaging application is from PE. Table 1.1 illustrates the total waste generated in peninsular Malaysia.

Table 1.1: Generation of total solid waste in Peninsular Malaysia (Johari et al., 2014)

Year	Total amount (Tonnes per day)
2010	23 000
2012	25 000
2020	30 000

The average amount of solid waste generated in Malaysia is 0.5-0.8 kg/person/day (Kathirvale et al., 2003). Figure 1.1 shows the components of solid waste composition in

Peninsular Malaysia. Plastics contributes about 11% from the component of solid waste and indicates the high amount waste disposal based on statistic generation of total solid waste in Peninsular Malaysia.

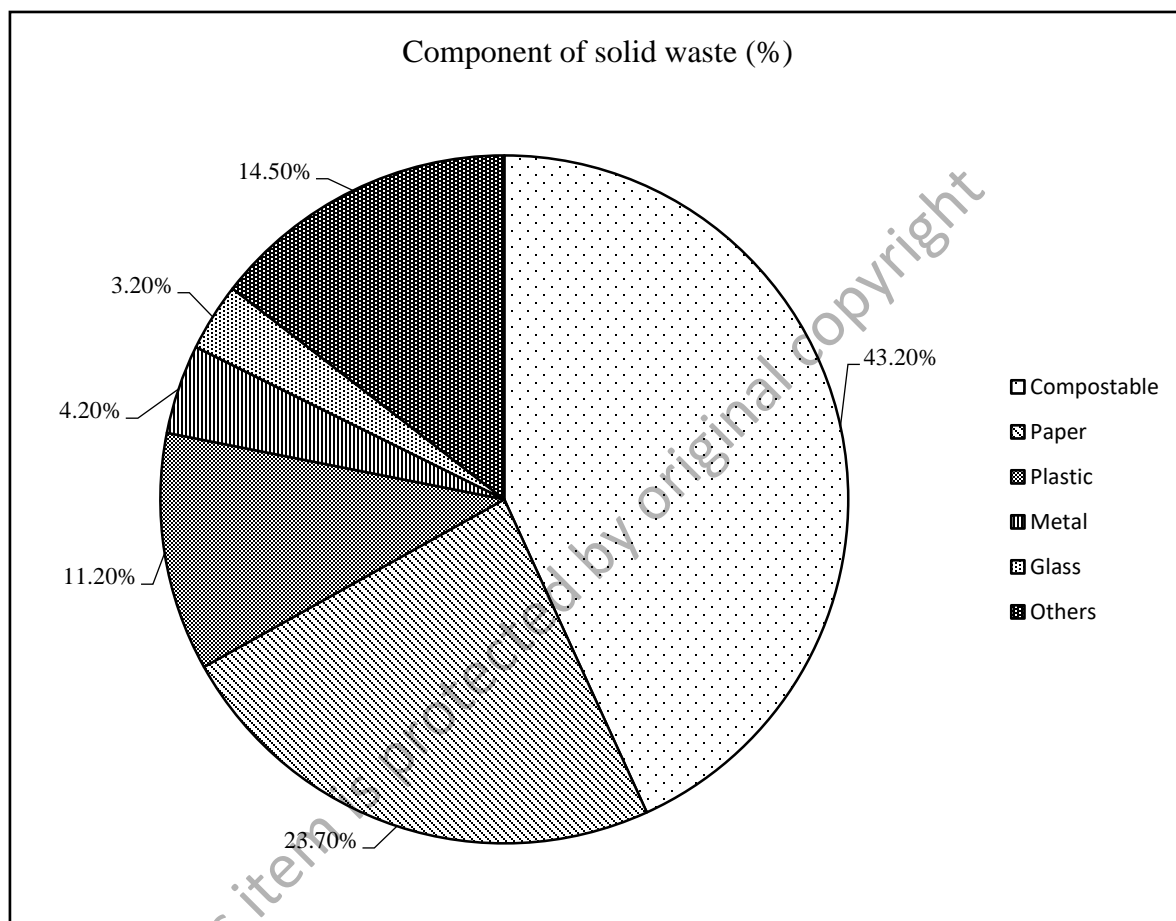


Figure 1.1: Component of solid waste (%) (Pusat Perbadanan dan Pengurusan Sisa Pepejal awam, 2014).

PE is non-degradable polymer and takes hundred years to completely degrade and highly resistant due to its chemical and environmental degradation (Otake et al., 1995). This is due to the properties of PE that has high molecular weight and cannot be degraded since it comprises of many branches that hardly to break down (Kathirvale et al., 2003). One of the

alternative accelerates the biodegradation of PE is the incorporation of natural filler into the polymer matrices.

Natural fibres that have been applied in non-degradable matrices are (i) leaf: sisal, pineapple leaf fibre (PALF), and henequen; (ii) bast: flax, ramie, kenaf/mesta, hemp and jute; (iii) seed: cotton; (iv) fruit: coconut husk, i.e., coir (Mohantya et al., 2000). Moreover, natural fibres are lignocellulosic in nature. Lignocellulosic materials are the most abundant renewable biomaterial of photosynthesis on earth. The lignocellulosic materials are widely distributed in the biosphere in the form of trees (wood), plants and crops. In lignocellulosic systems, the amount of cellulose, can vary depending on the species and age of the plant/species. Therefore, all of the natural fibres are hydrophilic in nature; their moisture content reaches 8 to 12.6% (Bledzki et al., 1996). Rambutan peel waste is one type of natural filler and readily abundant but does not have any market value. Therefore, the utilization of rambutan peel waste as natural filler in the polyolefins will provide a new application route to produce a low cost and degradable polymer. Rambutan peel waste as a lignocellulosic natural filler are degraded biologically. Biodegradation by the high molecular weight cellulose weakens the lignocellulosic cell wall and usually photochemical degradation by ultraviolet light occurs when lignocellulosics are exposed to outside exposure (Mohantya et al., 2000). Moreover, the utilization of rambutan peel waste can bring economic benefit and reduced the environmental problem impact. In recent years, the use of natural filler in polyolefins has gained great interest among researchers and industries due to some advantage of natural filler compared to mineral filler (e.g., kaolin, mica, talc, and calcium carbonate) such as low density, biodegradable, ecofriendly and renewable (Chun et al., 2013b; Salmah et al., 2013).

In general, the poor adhesion between the natural filler and the hydrophobic polyolefins is the main drawback to produce biocomposites. However, compatibilizer can be used to modify the hydrophilic natural filler, which can improve filler dispersion, wettability and filler-matrix adhesion. Previously, it was found that the used of used citric acid as a compatibilizer (Ning et al., 2007), glutaaldehyde as a crosslinking agent (Chan et al., 2013) and acrylic acid as chemical modification (Yeng et al., 2013) enhanced the properties of biocomposites. In current study, adipic acid (AA) was used as a compatibilizer. This compatibilizer was first used to compatibilize the polyolefin and rambutan peel. There is a lack of study in the utilization of adipic acid as a compatibilizer in polyolefin.

Apart from incorporating natural filler into non-degradable polymer, another approach to induce biodegradation of polyethylene is adding pro-oxidant into the polyolefin. Pro-oxidants are normally used for initiation of degradation including transition of complexes metal ions that can accelerate the oxidative process, consequently reducing the molecular weight of the polymer on the level that biodegradation takes place. It will act as initiator for the oxidation of the polyolefin. According to Reddy et al. (2009), the smaller segment of polymer chain can become nutrient for microorganism. Moreover, cobalt stearate (CS) has contributed the highest degradability to LDPE compared to the other cobalt carboxylates namely palmitate and laurate (Roy et al., 2007). Therefore, cobalt stearate has been applied as pro-oxidant in the present study.

## 1.2 Problem Statement

In recent years, the disposal of plastics especially those used in packaging, shows a serious challenge to waste management. Many efforts have been done on recycling the polyethylene including mechanical recycling, feedstock recycling and energy recovery. In order to reduce the waste problem of plastic, a replacement is needed. It is well known that polyethylene is highly resistant to biodegradation. One way by which the biodegradation of polyethylene can be accelerated is by the incorporation of a biodegradable filler such as fruit peel waste into the polymer matrix.

The incorporation of polyolefin with rambutan peel waste is necessary to produce a degradable plastic materials and hence resolve the waste problem from non-degradable plastic. However, one of the challenges is the compatibilization of rambutan peel waste and PE. Rambutan peel is hydrophilic material due to the presence of hydroxyl functional group. In contrast, PE is hydrophobic due to its hydrocarbon structure. In order to improve the properties of the biocomposites, compatibilizer is needed to compatibilize both materials.

The degradability of the polyolefin/biocomposites was always an issue among the researches. They claim that polyolefin that can be degraded during the degradation but the non-degradable component is still remain. Therefore, in order to overcome this issue, the potential of cobalt stearate as pro-oxidant in the biocomposites was investigated in the study.

### **1.3 Objectives of Study**

1. To evaluate the effect of rambutan peels flour (RPF) loading on LLDPE/RPF composites properties.
2. To study the effect of adipic acid as compatibilizer on the properties and degradability of LLDPE/RPF composites.
3. To investigate the effect of cobalt stearate as a pro-oxidant on the degradability of LLDPE/RPF-AA composites by natural weathering and soil burial.

### **1.4 Scope of Study**

This study focused on the development of degradable polymers by incorporating RPF into the LLDPE. The addition of adipic acid into RPF filler helps in dispersing RPF homogeneously within the LLDPE matrix. The composites are subjected to tensile, morphological and thermal properties testing. Then, the effect of natural weathering and soil burial test to the composites are investigated to determine their degradability. Other than that, the addition of cobalt stearate is aimed to accelerate the degradation process by accelerating the oxidation of LLDPE.

### **1.5 Thesis Organization**

This thesis contains five chapters with chapter one introduces briefly the scope of the thesis including the introduction about the research background, problem statement and